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17.2 Thermochemical Equations 1. Make the following conversions:  
a. 444 cal to joules =  $1.86 \times 10^3$  J  
b. 1.8 kJ to joules =  $1.8 \times 10^3$  J  
c. 0.45 kJ to calories =  $1.1 \times 10^2$  cal  
2.

Classify each of these processes as endothermic or exothermic: a. condensing steam ...

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SECTION 17.1 THE

FLOW OF

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WORK (pages 505–510)

This section explains

the relationship

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heat, and distinguishes between heat capacity and specific heat.

Energy

Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that

## **SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)**

SECTION 17.1 THE  
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WORK(pages505-510)

This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. ~ Energy Transformations(page505) 1.

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## Thermochemistry

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Thermochemistry I

Thermochemistry A.

“Thermo” refers to  
heat B.

Thermochemistry: the  
study of the transfers  
of energy as heat in  
chemical reactions and  
physical changes.

Energy is either gained  
or lost II Temperature

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and heat are related,  
but not identical A.

Heat 1. Heat is a form  
of energy 2.

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